

Name: \_\_\_\_\_ Date: \_\_\_\_\_

## Answer Key: The Secret Lego Bricks of Life: Organic Chemistry for Kindergartners

How do tiny invisible hands hold everything together? Kids will design and predict how carbon 'hooks' create the world around them.

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**1. Imagine Carbon is a friendly octopus. How many 'tentacles' or hands does he have to hold onto his friends?**

**Answer:** C) 4 hands

Carbon atoms have four valence electrons, meaning they like to make four bonds, or 'hands,' to connect with other atoms.

**2. True or False: Carbon atoms can hold hands with other Carbon atoms to make a very long parade line.**

**Answer:** A) True

This is called catenation! Carbon is special because it can form long, stable chains to build complex things like your toys or your skin.

**3. If we want to build a 'ring' of Carbon atoms to make a shape like a crown, what is the smallest number of friends we need to close the circle?**

**Answer:** C) 3 friends

Just like a triangle is the simplest closed shape, organic molecules need at least three carbons to form a ring structure.

**4. If Carbon and Hydrogen joined together to make a 'necklace' that smells like a lemon, what are they building?**

**Answer:** B) A molecule

A molecule is a group of atoms bonded together. When carbon forms specific patterns, it creates the scents we find in nature.

**5. True or False: Every single thing that is alive, like a puppy or a tree, is made using Carbon building blocks.**

**Answer:** A) True

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Organic chemistry is the study of life! Carbon is the foundational element for all living organisms on Earth.

**6. Carbon is very strong. If we squeeze Carbon atoms together very, very hard for a long time, they can turn into a shiny \_\_\_\_.**

**Answer:** C) Diamond

Diamonds are made entirely of carbon atoms arranged in a very strong, repeating crystal pattern.

**7. Sometimes Carbon atoms use TWO hands to hold onto ONE friend. What do scientists call this 'double grip'?**

**Answer:** A) A double bond

In chemistry, when two atoms share four electrons (two pairs), it is called a double bond, which makes the connection stronger and changes the shape.

**8. True or False: You can turn a piece of soft plastic into a different shape because the Carbon chains are sliding past each other like cooked spaghetti.**

**Answer:** A) True

Polymers (like plastics) are long carbon chains. Their flexibility often comes from how these long chains interact and move.

**9. We breathe out a gas that has one Carbon and two Oxygens. It is what plants 'eat' to grow. What is it?**

**Answer:** B) Carbon Dioxide

Carbon dioxide (CO<sub>2</sub>) is a simple organic-related molecule where carbon connects with oxygen, showing how carbon moves through the air and living things.

**10. If you were a chemist and you wanted to build a molecule that was very STIFF and didn't move, which shape would you use?**

**Answer:** C) A criss-cross cage

Cross-linking or 'cage' structures (like in graphene or diamond) create rigid materials, while straight lines are often more flexible.