

Name: \_\_\_\_\_ Date: \_\_\_\_\_

## Answer Key: Heat Hero Challenges for Kindergarten

Can your students predict where the heat will hide? Young scientists will analyze heat flow and material changes in these challenging scenarios.

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**1. If you leave a cold popsicle and a warm cookie on a table in a sunny room, what will eventually happen to both of them?**

**Answer:** B) They will both become the same temperature as the room

Heat moves between objects and the air until everything reaches a state of balance, which is the room's temperature.

**2. If you have a hot cup of cocoa, the heat energy can travel through the air to reach your cold hands without you touching the cup.**

**Answer:** A) True

This is thermal radiation; heat can travel through space or air to reach nearby objects even without direct contact.

**3. You place a metal spoon and a plastic spoon into a warm bowl of soup. Which spoon will feel the hottest to touch after one minute?**

**Answer:** C) The metal spoon

Metal is an excellent conductor, meaning heat energy flows through it much faster than through plastic.

**4. Imagine you are wearing a black shirt and your friend is wearing a white shirt on a very sunny day. Why do you feel much hotter than your friend?**

**Answer:** A) Dark colors soak up more heat from sunlight

Dark objects absorb more radiant energy from the sun, which turns into thermal energy and makes the fabric hotter.

**5. When you put an ice cube into a glass of warm water, the 'coldness' moves from the ice into the water.**

**Answer:** B) False

Scientifically, 'cold' doesn't move. Instead, the heat energy from the warm water moves into the ice cube to melt it.

**6. If you hold a piece of chocolate in your hand, it melts. Where did the heat energy come from to make the chocolate change?**

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**Answer:** B) Your warm body

Your body has a higher temperature than the chocolate, so heat transfers from your skin to the chocolate, causing a phase change.

**7. A scientist wants to keep a cup of juice cold. Which material would be the BEST 'Heat Trapper' (insulator) to wrap around the cup?**

**Answer:** B) A thick wool sock

Wool is an insulator that slows down the transfer of heat from the warm air to the cold juice.

**8. If you rub your hands together very quickly, they will start to feel warm.**

**Answer:** A) True

Friction (rubbing) turns the energy of your movement into thermal energy, which makes your skin feel hot.

**9. You have two bowls of soup. One is very hot and one is just warm. Which one will lose its heat to the cool air faster?**

**Answer:** A) The very hot soup

The bigger the difference in temperature between the soup and the air, the faster the heat energy will move.

**10. If a bird wants to stay warm during a snowy day, why does it fluff up its feathers?**

**Answer:** B) To trap a layer of air that keeps its body heat in

Trapped air is an excellent insulator; fluffing feathers creates an air pocket that stops body heat from escaping into the cold.